

Total Synthesis of Palmarumycin CP₁ and (±)-Deoxypreussomerin A

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Since the isolation of the spiroketal MK3018 by Ogishi et al. in 1989¹ and the bis-ketal decalin preussomerin A by Weber et al. in 1990,² many structurally related fungal metabolites have been reported (Figure 1).³ In addition to general antitumor and antibiotic activities,³ selective inhibition of Ras farnesyl-protein transferase with an IC₅₀ of ≥ 1.2 μ M has been detected with these compounds.⁴

Recently, Krohn et al. reported the first semisynthetic generation of the palmarumycin spiroacetal framework by oxidative cyclization of a fungal metabolite.⁵ As a part of our program toward the synthesis and mechanistic study of epoxy ketone natural products,^{6,7} we have recently embarked on the total synthesis of diepoxin σ .⁸ In this paper, we describe our progress toward a general route for the construction of the spiroketal naphthodecalins and, specifically, the first total syntheses of palmarumycin CP₁ and (±)-deoxypreussomerin A. These natural products were isolated from *Coniothyrium palmarum*^{3f} and an unidentified ceolomycetes (MF 5916),⁴ respectively, and are representative for the structural features of many naturally occurring naphthoquinone dimers. Deoxypreussomerin A was also isolated (and named palmarumycin C₂) from *C. palmarum*, and the structure was further confirmed by X-ray analysis.^{3g} We envisioned that both compounds as well as ultimately diepoxin σ could be derived from *p*-quinone ketal **1**, which

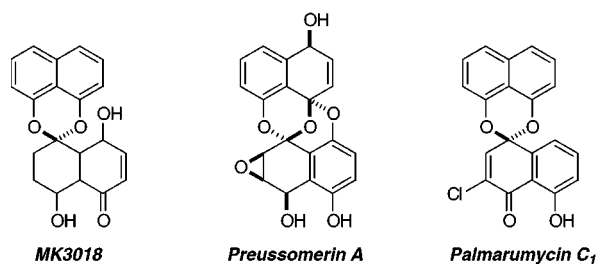
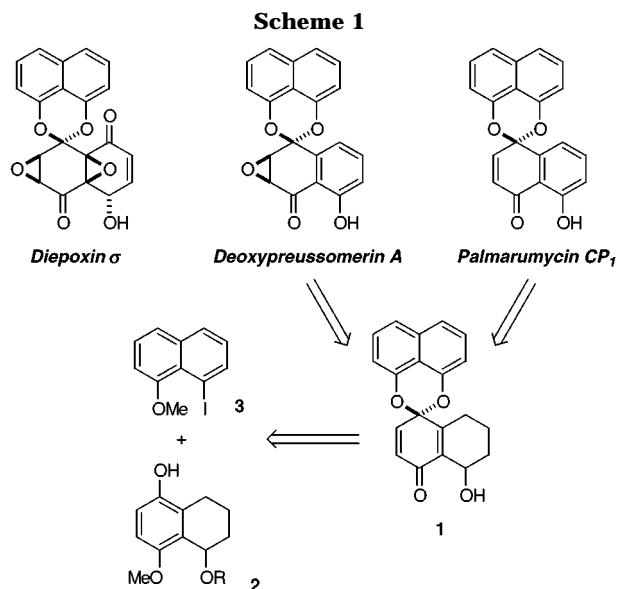


Figure 1. Representative spiroketal naphthodecalin natural products.



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should be available by Ullmann ether coupling of phenol **2** and iodide **3** followed by oxidative spirocyclization (Scheme 1).

Our first attempts for an Ullmann coupling⁹ between the readily available¹⁰ tetralone **4** and 8-iodo-1-methoxynaphthalene (**3**)¹¹ failed, possibly due to the lack of nucleophilicity of the deactivated tetralone under the standard Ullmann reaction conditions. After protection of the carbonyl group in **4** with ethylene glycol, the resulting acetal was now successfully converted to the biaryl ether **5** in 78% yield in refluxing pyridine in the presence of 0.2 equiv of Cu₂O and 1 equiv of K₂CO₃ (Scheme 2). When 1 equiv of Cu₂O was used, the coupling yield dropped to <60%.

O-Demethylation in the presence of the dioxolane protective group was not clean; however, stepwise cleavage of the acetal in aqueous acetone with TsOH followed by double demethylation with BBr₃ gave **7** in 95% overall yield. After considerable experimentation, we found that the oxidative spirocyclization of phenol **7** was best performed after reduction of the tetralone with LAH by treatment with PhI(OAc)₂ in trifluoroethanol for 10 min.^{6,8,12} The linchpin spiroketal intermediate **1** was thus obtained in high overall yield.

For the preparation of palmarumycin, spiroketal **1** was oxidized with Dess–Martin reagent,¹³ followed by aromati-

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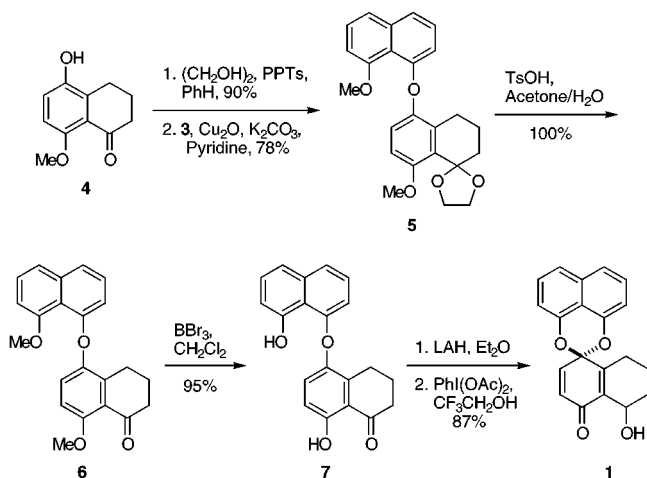
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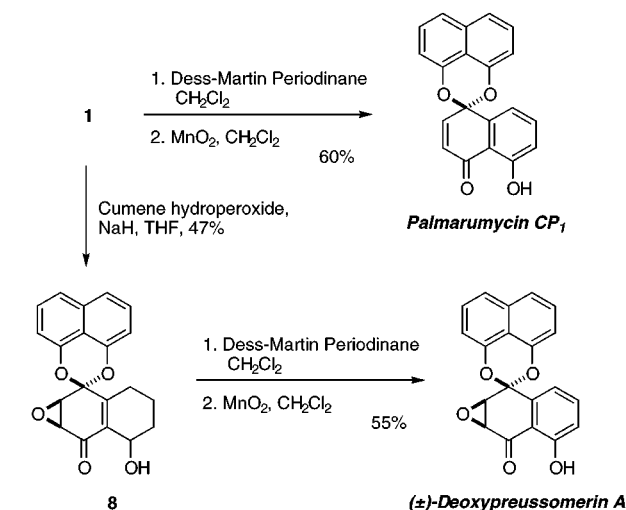
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Scheme 2



Scheme 3



zation of the resulting decalin dione with activated MnO_2 (Scheme 3). This two-step protocol provided significantly higher yields than simple exposure of **1** to an excess of MnO_2 , particularly since the product was difficult to remove from this reagent. Synthetic palmarumycin, which was spectroscopically in all aspects identical to the data reported for the fermented compound,¹⁴ was thus obtained in 35% overall yield in eight steps from tetralone **4**. The same strategy

provided (\pm)-deoxypreussomerin A after epoxidation of **1** with cumene hydroperoxide anion in THF in 15% overall yield based on a nine-step conversion from **4**.¹⁵

The structural features of many dimeric naphthoquinone natural products are deceptively simple. In particular, the actual introduction of the spiroketal moiety by dehydration of a *p*-quinone or transacetalization of a semiquinone precursor with 1,8-naphthalenediol has, in our hands, so far been unsuccessful. The use of the Ullmann ether coupling followed by oxidative spirocyclization, however, provides a high-yielding access to the intermediate **1**, which should be useful for the preparation of gram quantities of many spiroketal naphthodecalin metabolites. Specifically, we have achieved the first total syntheses of palmarumycin CP_1 and deoxypreussomerin A, two representative members of this class of natural products. Our work provides further support for the intermediacy of phenolic binaphthyl ethers related to **6** or **7** in the biosynthesis of naphthoquinone natural products, as first postulated by Krohn et al.^{3g,5} We are currently extending our biomimetic strategy toward the enantioselective synthesis of diepoxin σ .

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Supporting Information Available: Experimental details and characterization for **1**, **6**, **7**, palmarumycin CP_1 , and (\pm)-deoxypreussomerin A (16 pages).

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(14) Spectroscopic data for synthetic palmarumycin CP_1 : mp 170 °C dec; IR (neat) 3053, 1659, 1602, 1449, 1409, 1372, 1341, 1269, 1237, 1110, 1073, 942, 822, 746 cm^{-1} ; ^1H NMR (300 MHz, CDCl_3) δ 12.17 (s, 1 H, OH), 7.67 (t, 1 H, $J = 8.0$ Hz), 7.58 (d, 2 H, $J = 8.5$ Hz), 7.47 (t, 2 H, $J = 7.9$ Hz), 7.46 (d, 1 H, $J = 7.8$ Hz), 7.14 (dd, 1 H, $J = 8.2, 1.1$ Hz), 7.02 (d, 1 H, $J = 10.9$ Hz), 6.98 (d, 2 H, $J = 7.7$ Hz), 6.37 (d, 1 H, $J = 10.9$ Hz); ^{13}C NMR (75 MHz, CDCl_3) δ 188.8, 161.9, 147.2, 139.7, 138.8, 136.7, 134.2, 129.8, 127.7, 121.4, 119.7, 119.4, 113.8, 113.0, 109.9, 92.9; MS (EI) m/z (rel intensity) 316 (M^+ , 100), 288 (12), 287 (19), 259 (8), 175 (11), 114 (45), 88 (11), 63 (9); HRMS (EI) calcd for $\text{C}_{20}\text{H}_{12}\text{O}_4$ 316.0736, found 316.0730.

(15) Spectroscopic data for synthetic (\pm)-deoxypreussomerin A (palmarumycin C_2): mp 200–201 °C; IR (neat) 3050, 1651, 1605, 1455, 1409, 1380, 1330, 1266, 1239, 1173, 1110, 1061, 963, 920, 878, 820, 809, 759, 720 cm^{-1} ; ^1H NMR (300 MHz, CDCl_3) δ 11.37 (s, 1 H, OH), 7.65 (t, 1 H, $J = 8.0$ Hz), 7.60 (d, 1 H, $J = 8.6$ Hz), 7.57 (d, 1 H, $J = 8.0$ Hz), 7.53 (t, 1 H, $J = 8.3$ Hz), 7.45 (t, 1 H, $J = 7.4$ Hz), 7.44 (d, 1 H, $J = 7.9$ Hz), 7.19 (dd, 1 H, $J = 7.6, 0.8$ Hz), 7.14 (dd, 1 H, $J = 8.6, 0.8$ Hz), 6.92 (dd, 1 H, $J = 7.6, 0.7$ Hz), 4.09 (d, 1 H, $J = 4.1$ Hz), 3.68 (d, 1 H, $J = 3.9$ Hz); ^{13}C NMR (75 MHz, CDCl_3) δ 196.6, 161.9, 146.9, 146.7, 137.8, 136.9, 134.2, 127.9, 127.7, 121.5, 121.4, 120.1, 119.1, 112.8, 112.3, 110.2, 109.4, 96.0, 53.3; MS (EI) m/z (rel intensity) 332 (M^+ , 100), 316 (28), 303 (11), 287 (19), 173 (15), 145 (23), 132 (12), 114 (27), 89 (13), 74 (14), 63 (12), 57 (7); HRMS (EI) calcd for $\text{C}_{20}\text{H}_{12}\text{O}_5$ 332.0685, found 332.0688.